

# **Mechanochemical Modification of Lignin and Application of the Modified Lignin for Polymer Materials**

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# Significance

**Petroleum-based products have big issues**



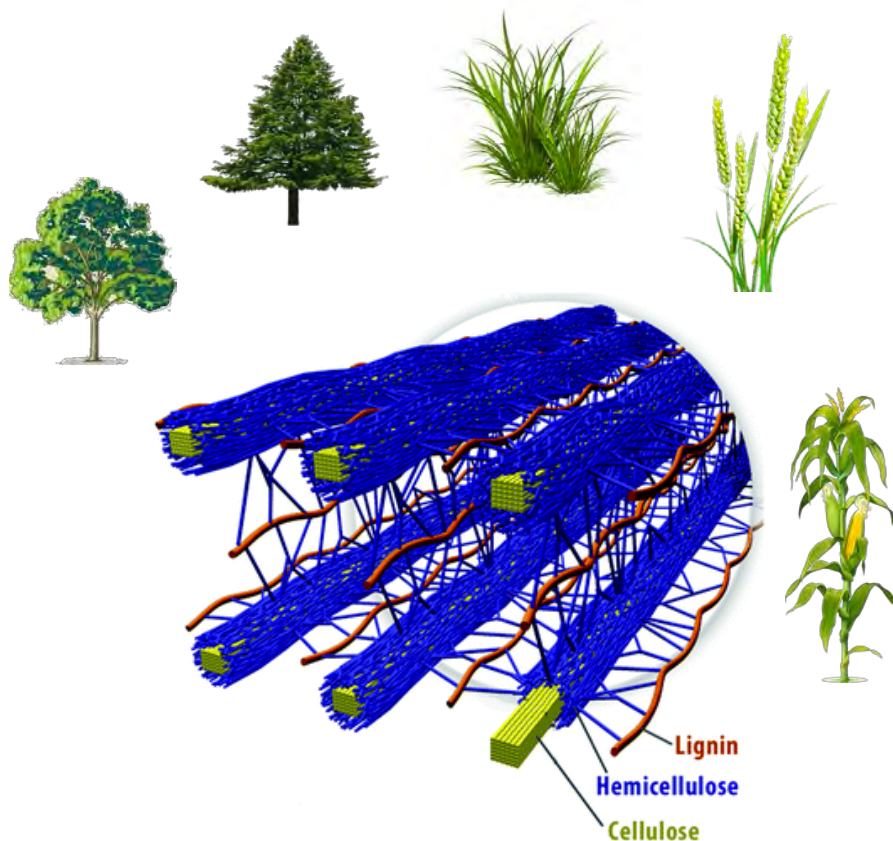
**A shift to biobased products**



## **Lignocellulosic biomass**

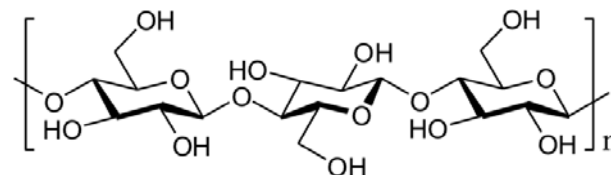
- ✧ most abundant renewable resource
- ✧ annual yield: 200 billion tons

# Lignocellulosic Biomass



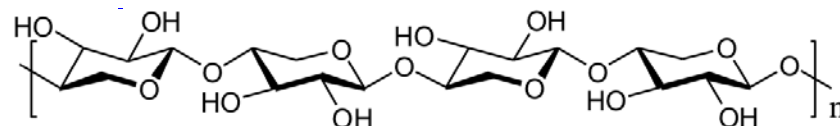
## Cellulose 40-45%

- Crystalline
- Linear polysaccharide of



## Hemicellulose 25-35%

- Amorphous
- Branched polysaccharide of

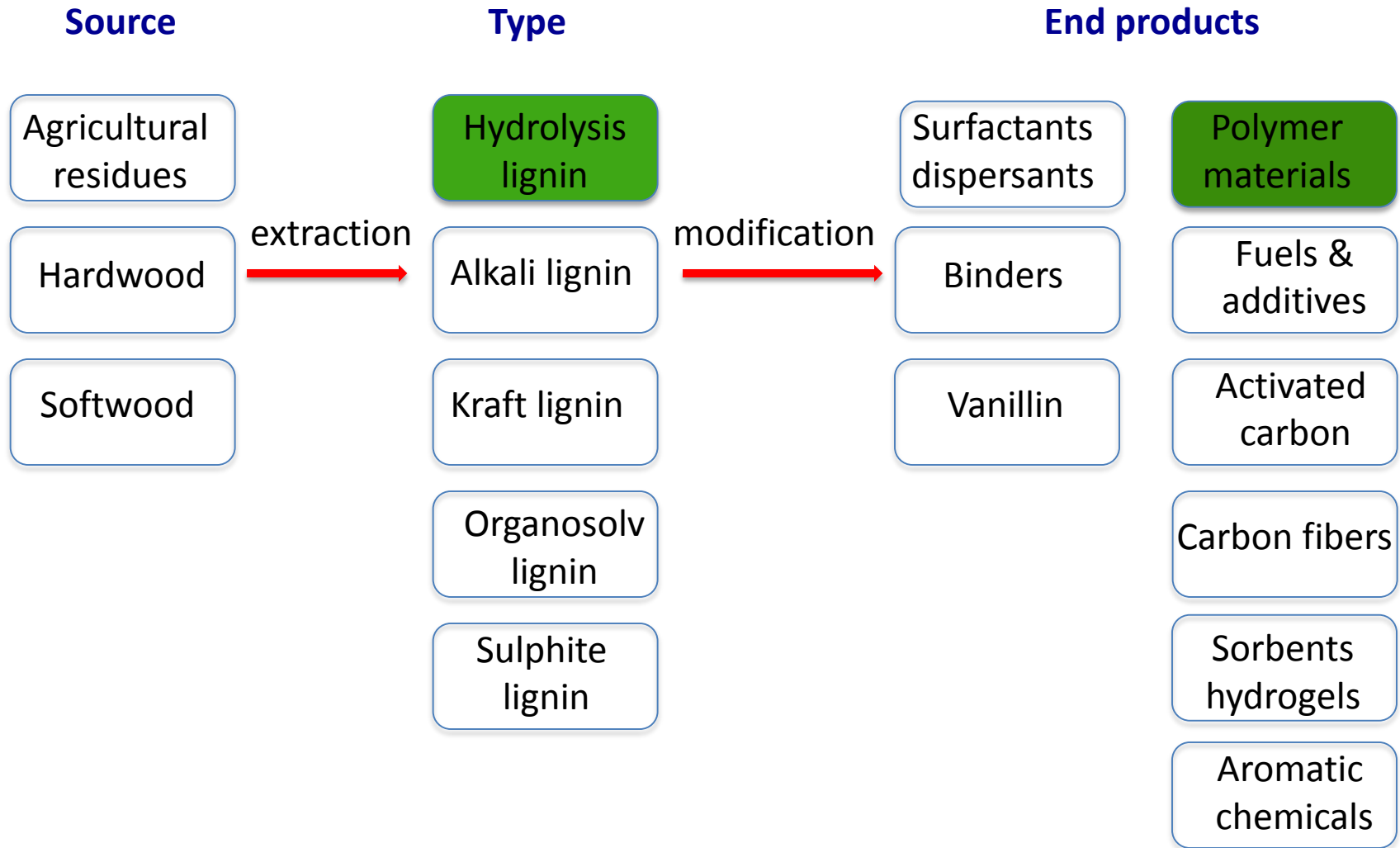


## Lignin 10-30%

- Amorphous
- Highly branched aromatic polymer
- Phenylpropanols

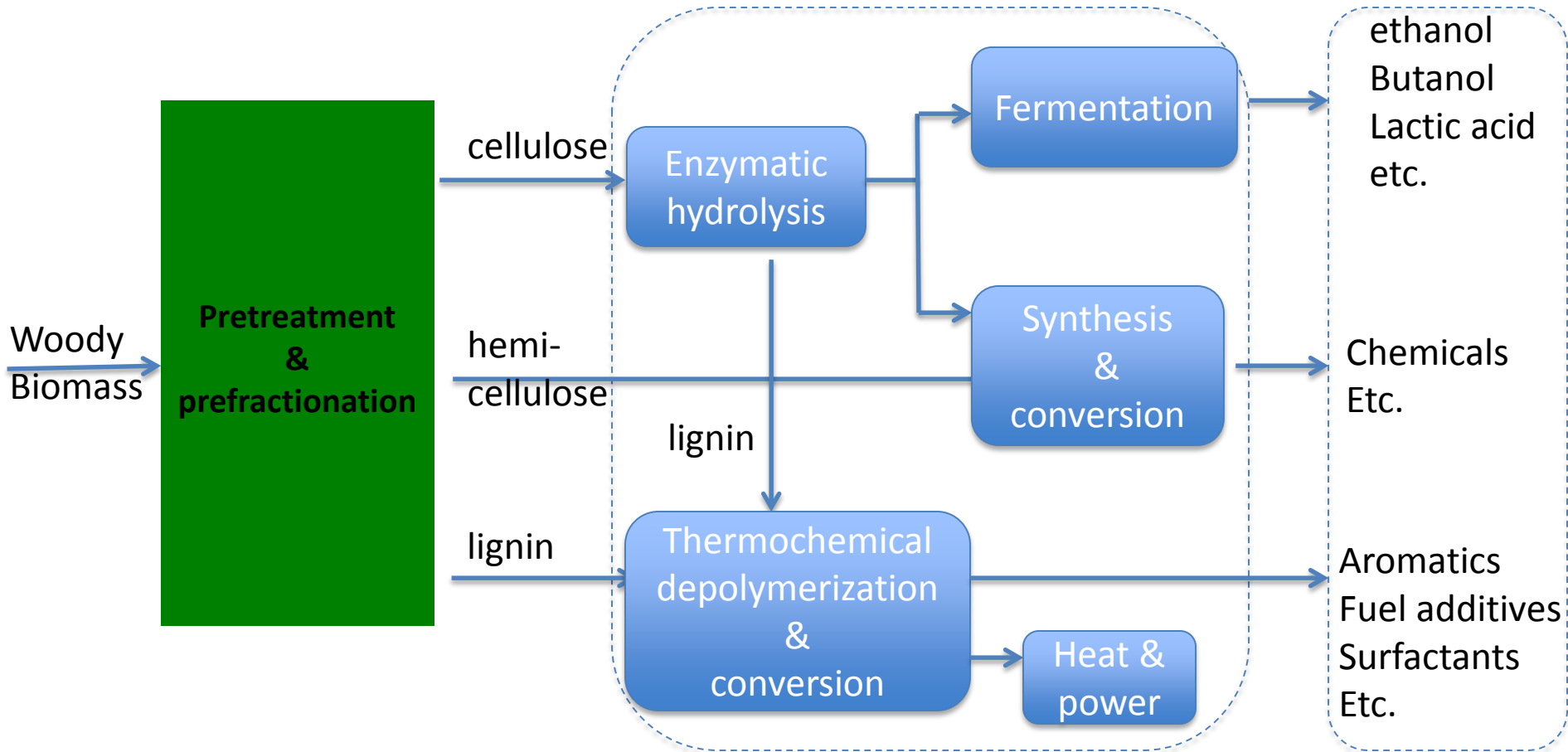
A Brandt, J Gräsvik, JP Hallett, T Welton.  
Green chem., 2013, 15, 550-583.

# Sources, types and potential products and applications of lignin



# Scheme of a possible biorefinery

**Fuels  
chemicals  
Materials**



# Conversion of Lignin to bioproducts

## Lignin

Pyrolysis, thermolysis, hydrogenolysis,  
gasification, hydrolysis, oxidation

Methylation, butylation, glycidylation,  
amination, acetylation

### depolymerization

### Chemical modification

#### Chemicals

(phenols, vanillin,  
syringaldehyde,  
aliphatics, etc.)

#### Fuel

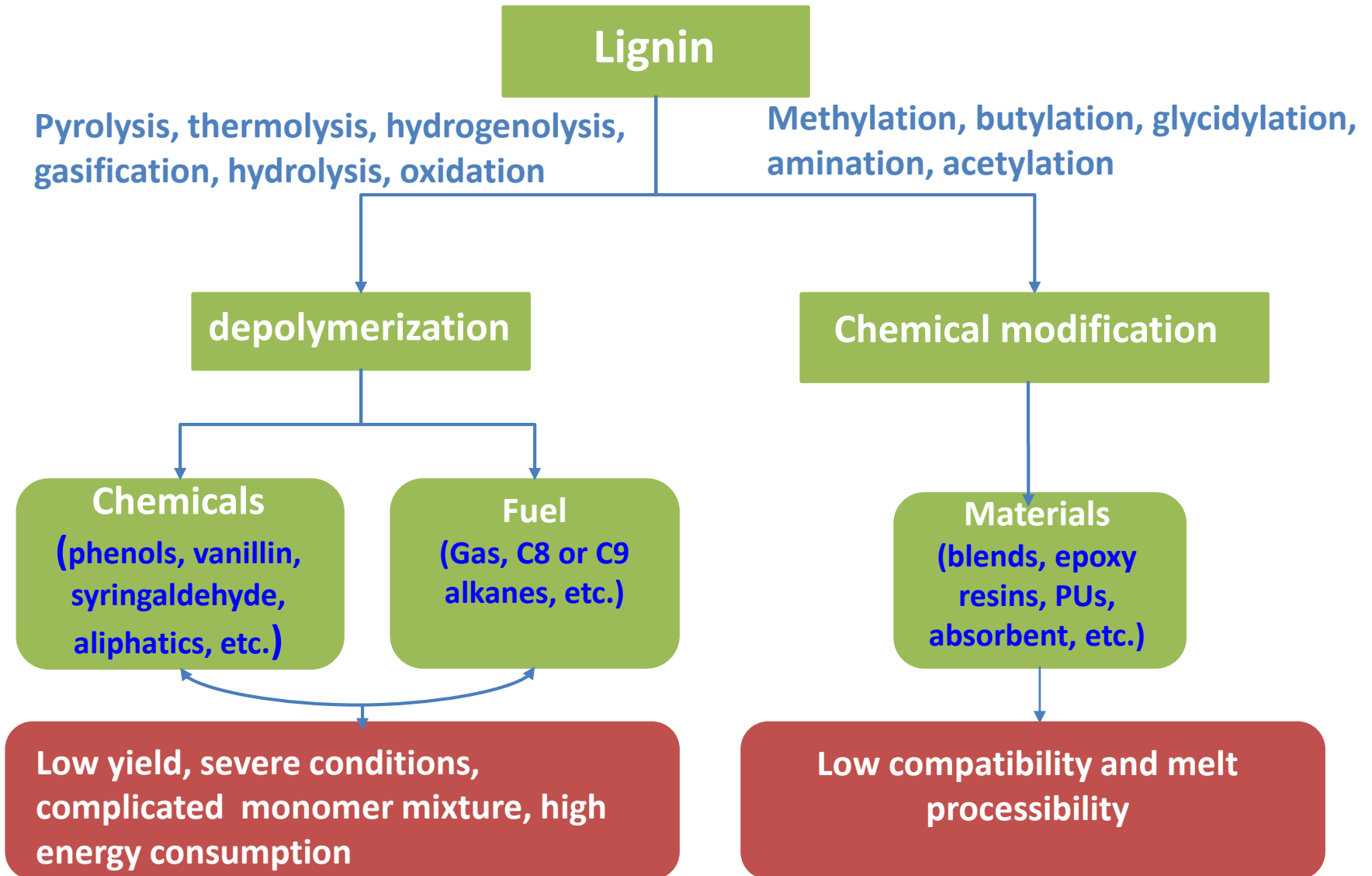
(Gas, C8 or C9  
alkanes, etc.)

#### Materials

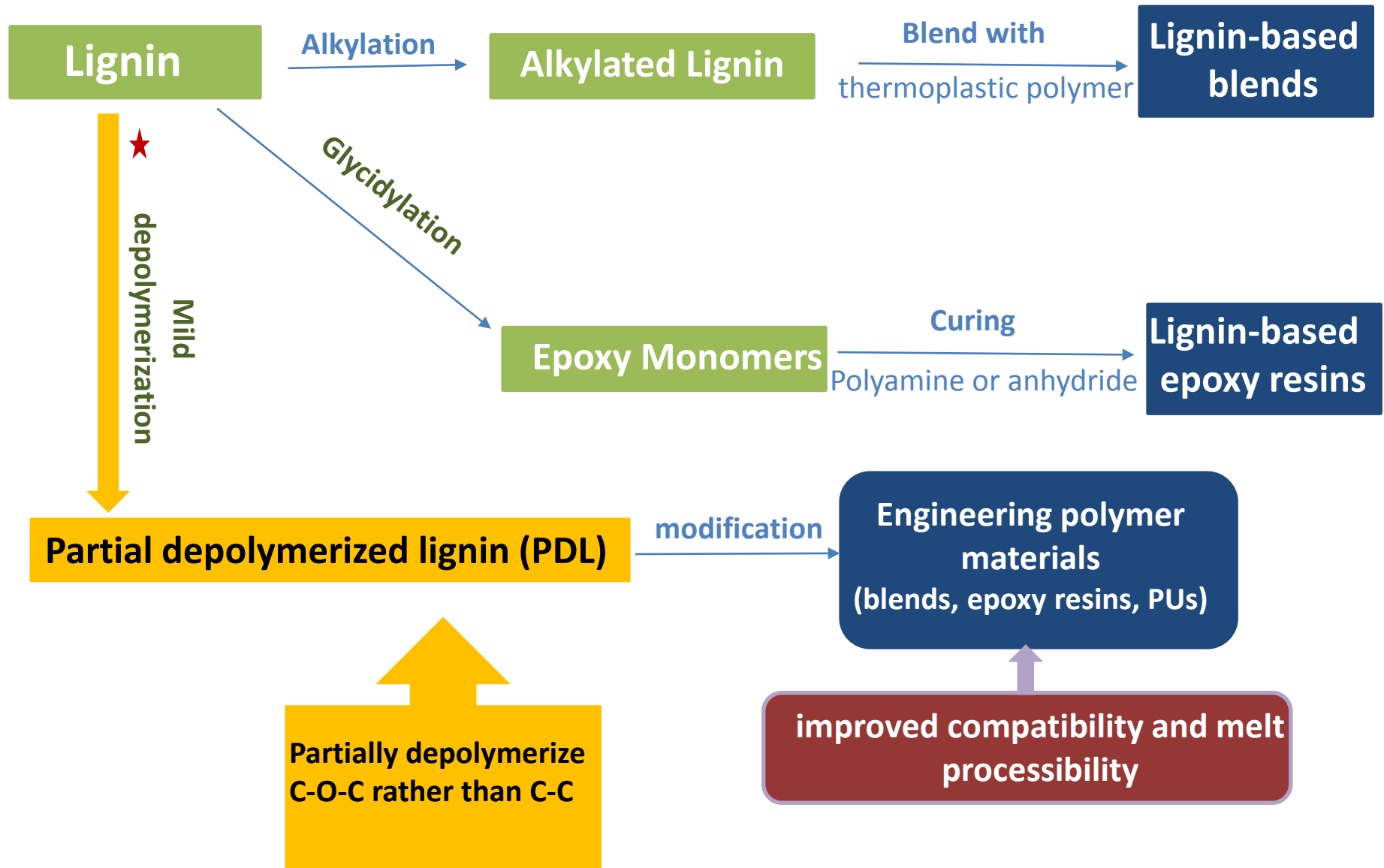
(blends, epoxy  
resins, PUs,  
absorbent, etc.)

Low yield, severe conditions,  
complicated monomer mixture, high  
energy consumption

Low compatibility and melt  
processibility



# Use of lignin for materials - Methods



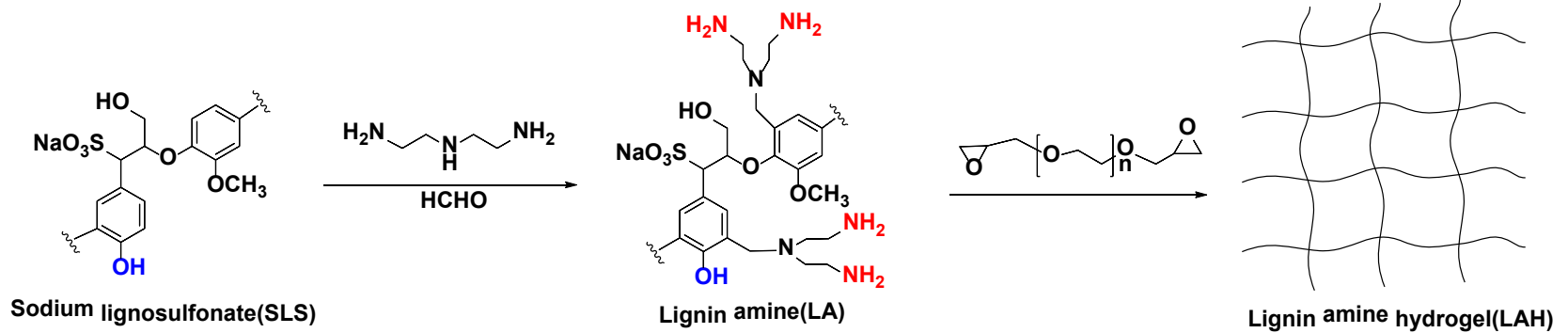
# Hydrogel made from lignosulfonate



Mannich reaction

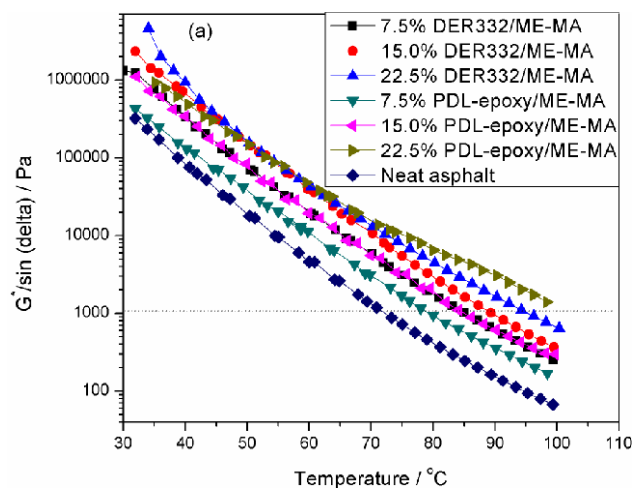
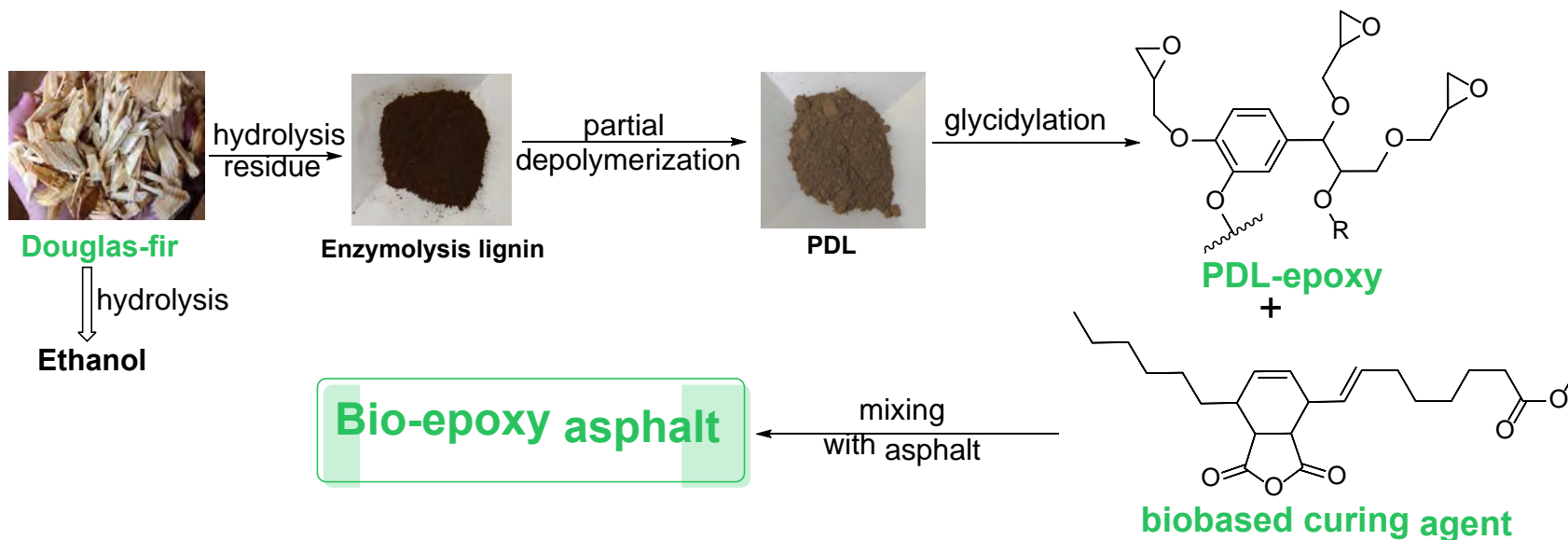


Crosslinking reaction





# Lignin-derived epoxy modified asphalt



ACS Sustainable Chemistry and Engineering 2016

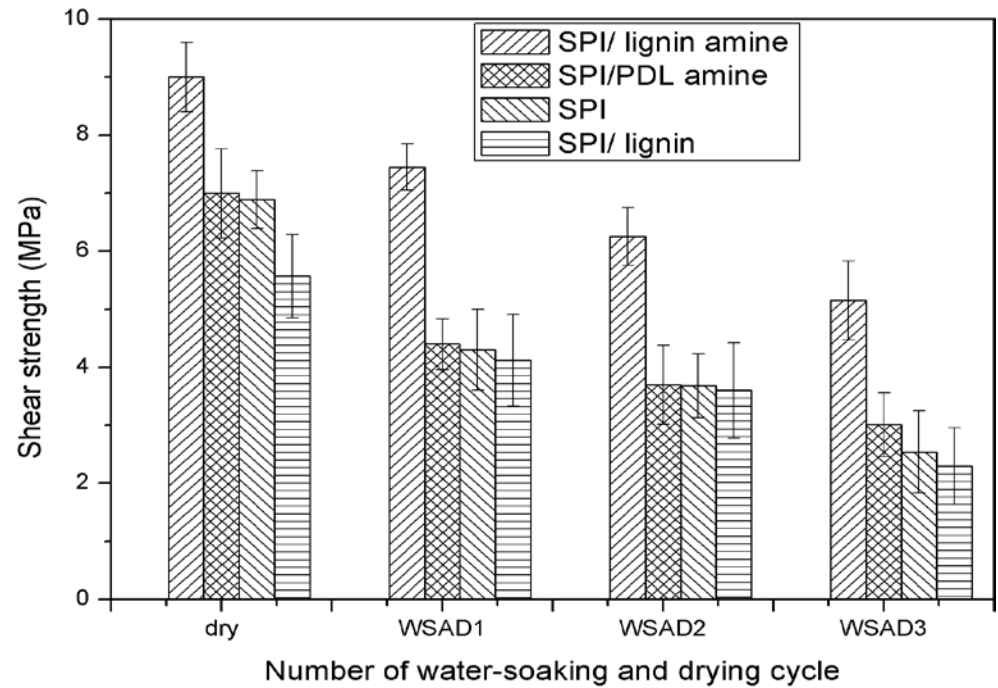
# Lignin modified soy protein adhesives for wood composites



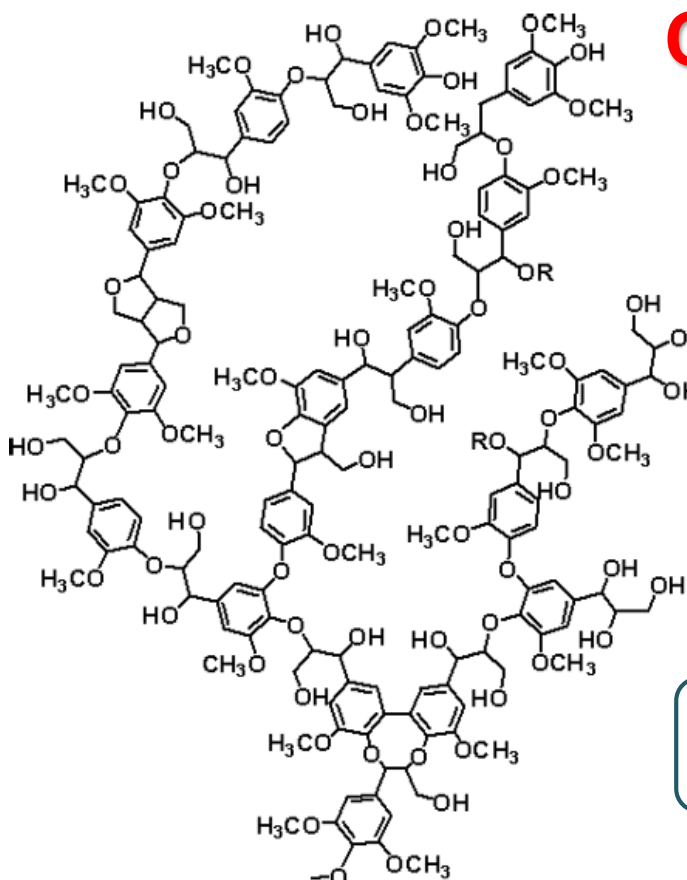
Lignin amine



Soy protein  
adhesive



# Challenges of lignin modification

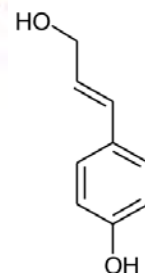


<https://www.e-education.psu.edu/egee439/node/665>

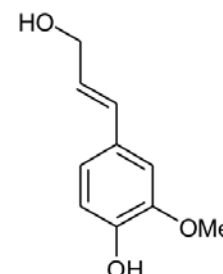


abundant hydroxyls

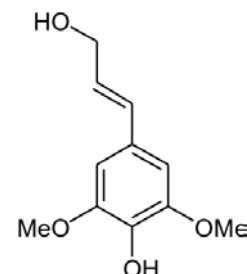
Potential feedstock: value-added chemical compounds



*coumaryl alcohol*



*coniferyl alcohol*



*syringyl alcohol*

Highly branched structure  
Strong intramolecular force  
Low solubility in solvents

Low accessibility of  
hydroxyls (OH)

Low reaction efficiency  
and conversion

**Need** a simple and green method for lignin modification

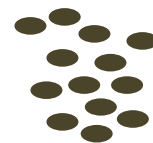
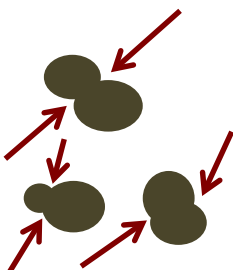
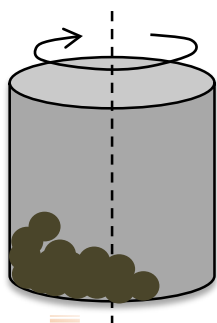
# Mechanochemistry

## History

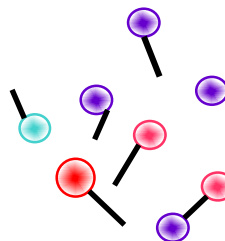
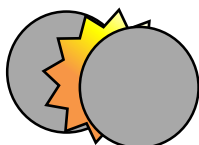
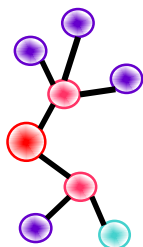


## Merits

- ✧ Initiate reaction in the absence of solvent
- ✧ Reduce by-products and toxic wastes
- ✧ Reduce reaction time (energy savings)

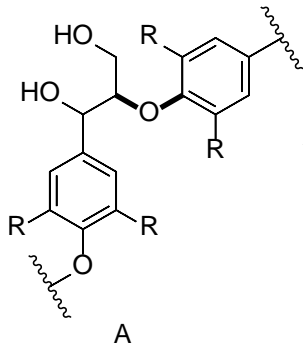


**Particle size**



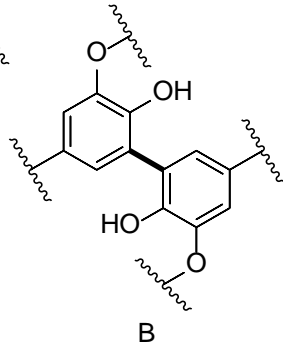
**Chemical bonds**

# Major linkages in lignin



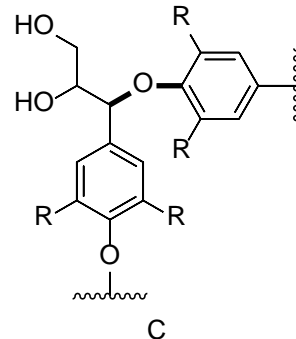
46% (SW)  
60% (HW)

$\beta$ -O-4



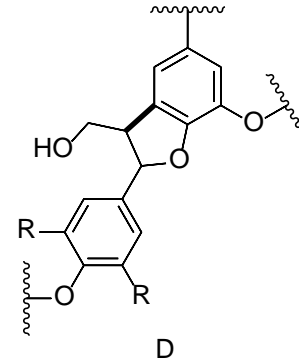
9.5-11% (SW)  
4.5% (HW)

5-5



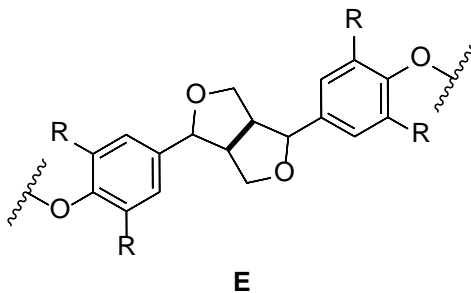
6-8% (SW)  
6-8% (HW)

$\alpha$ -O-4



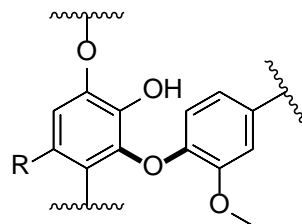
9-12% (SW)  
6% (HW)

$\beta$ -5



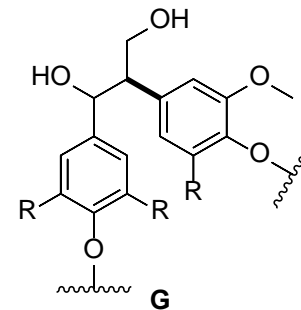
2% (SW)  
3% (HW)

$\beta$ - $\beta$



3.5-4% (SW)  
6.5% (HW)

4-O-5



7% (SW)  
7% (HW)

$\beta$ -1

# Mechanochemistry in lignocellulosic biomass

Mechanical energy



Initiate reaction  
Structural change

## Isolation of lignin from wood and pulp

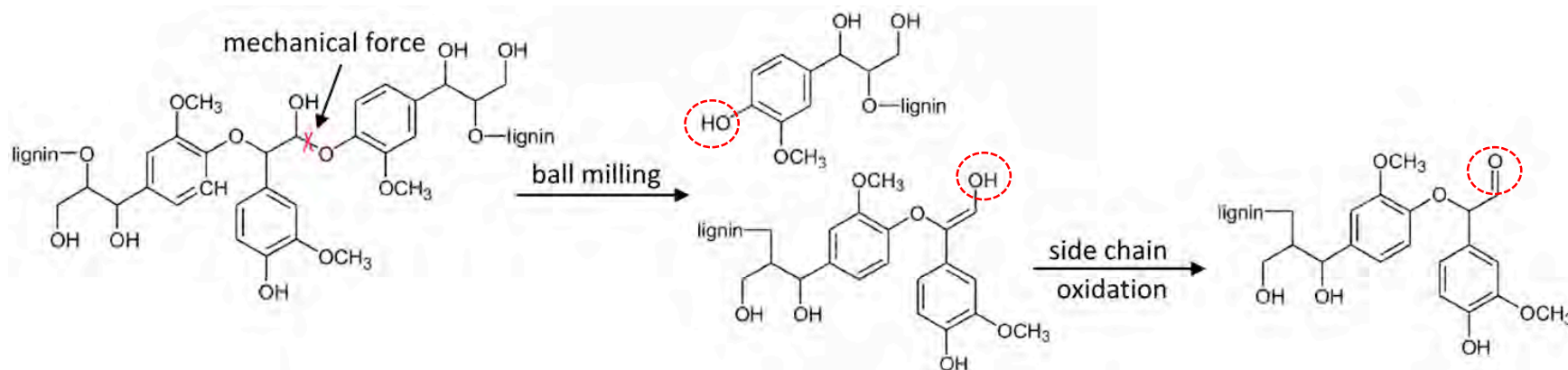
### Free radicals

undergo reaction with  
radical scavenger

Ball milling

Yield  
Molecular weight  
Structure

## Cleavage process of $\beta$ -O-4 linkages in lignin



# Lignin polymer materials for engineering application

- **Lignin as feedstock for thermoplastics**

- Mechanochemical modification of lignin  
Transesterification between lignin and fatty oils
- Modified lignin-based polymer blends

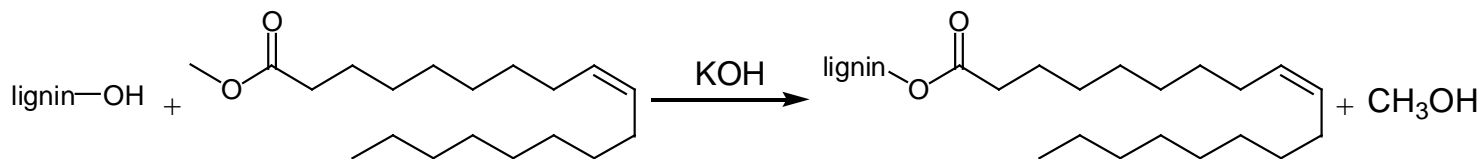
- **Lignin as feedstock for thermosets**

- Mechanochemical modification of lignin  
Esterification between lignin and anhydrides
- Modified lignin-derived cured epoxy resins

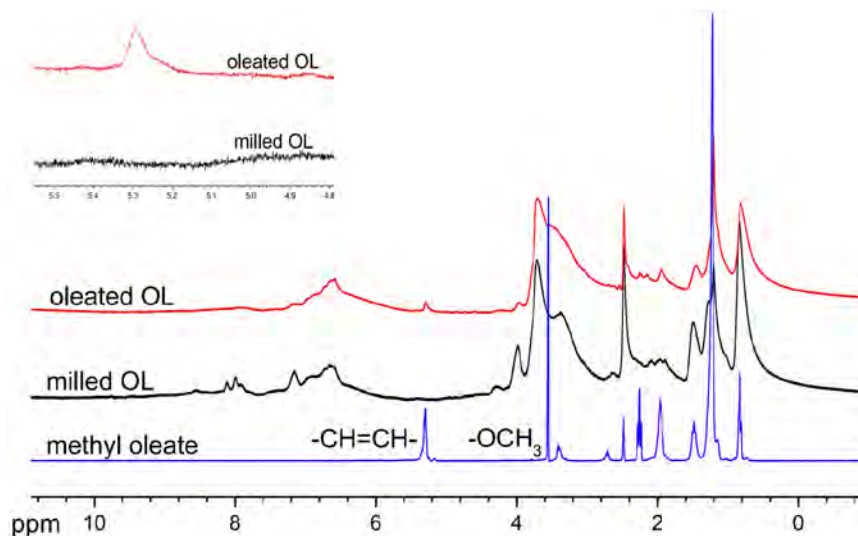




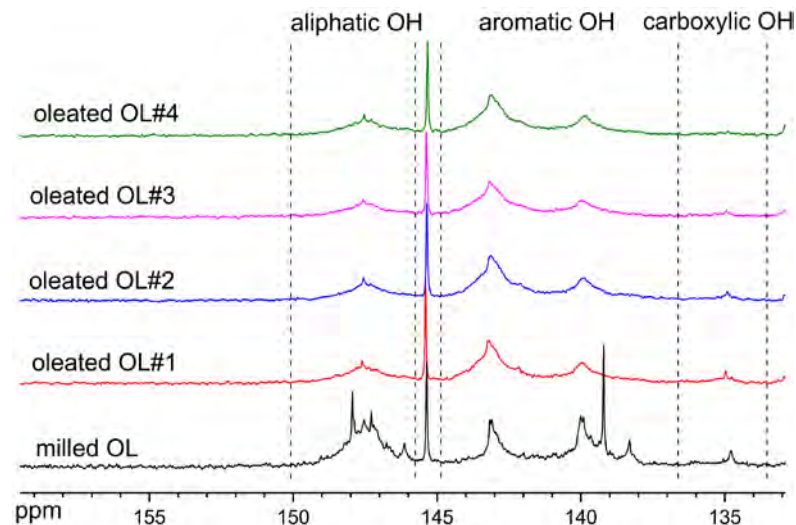
## Transesterification between lignin and fatty oils



### <sup>1</sup>H NMR spectra



### <sup>31</sup>P NMR spectra



## 31 P NMR Spectra of oleated *organosolv lignin* (OL)



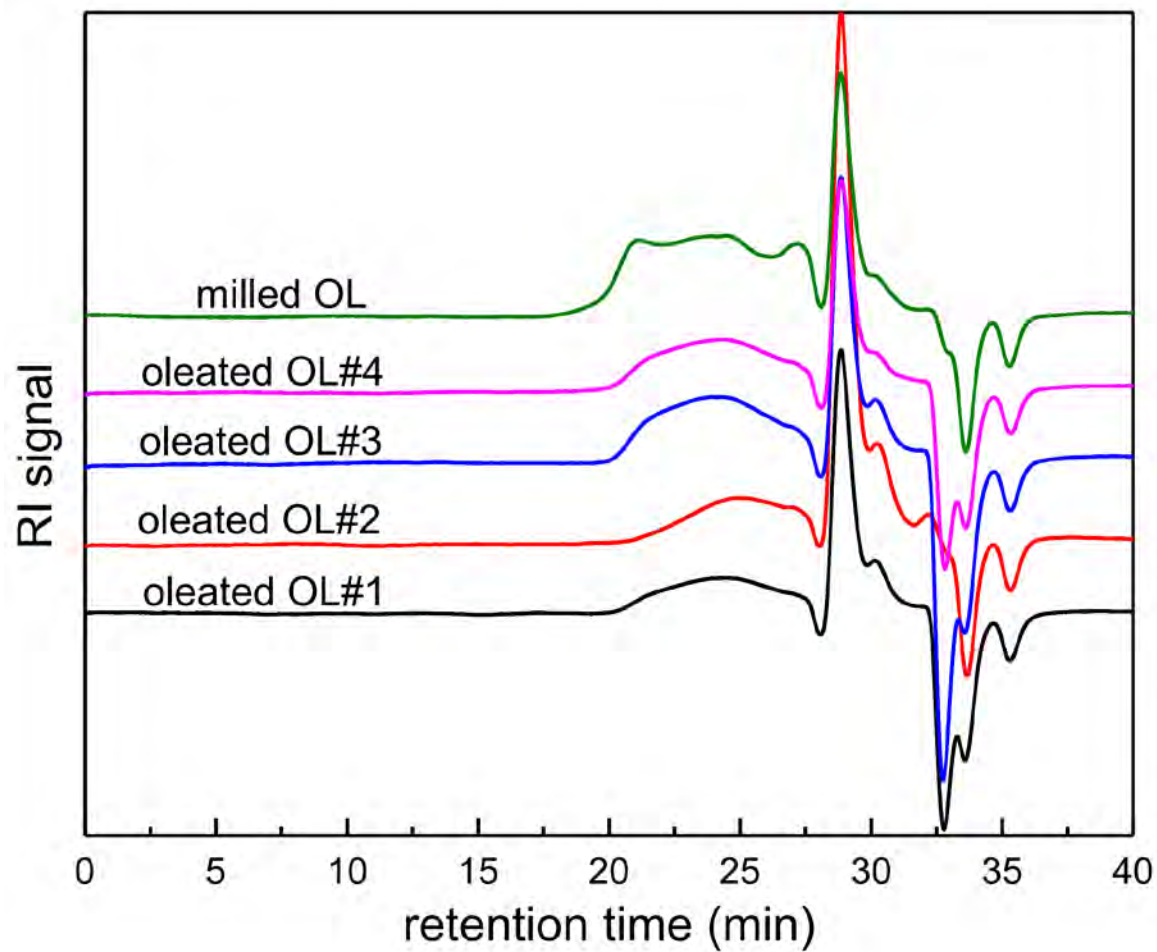
**Effects of oleation stoichiometry on conversion and hydroxyl value of modified lignin**

Sample	$n_{\text{L-OH}}:n_{\text{Mo}}$ (molar ratio)	Conversion (%)	Hydroxyl value (mmol/g)		
			Aliphatic OH	Aromatic OH	Total
milled OL	/	/	2.35	2.73	5.08
oleated OL#1	1:0.5	25.02	0.70	2.19	2.89
oleated OL#2	1:0.6	25.46	0.64	2.22	2.86
oleated OL#3	1:0.7	24.43	0.59	2.34	2.93
oleated OL#4	1:0.8	22.72	0.72	2.33	3.05

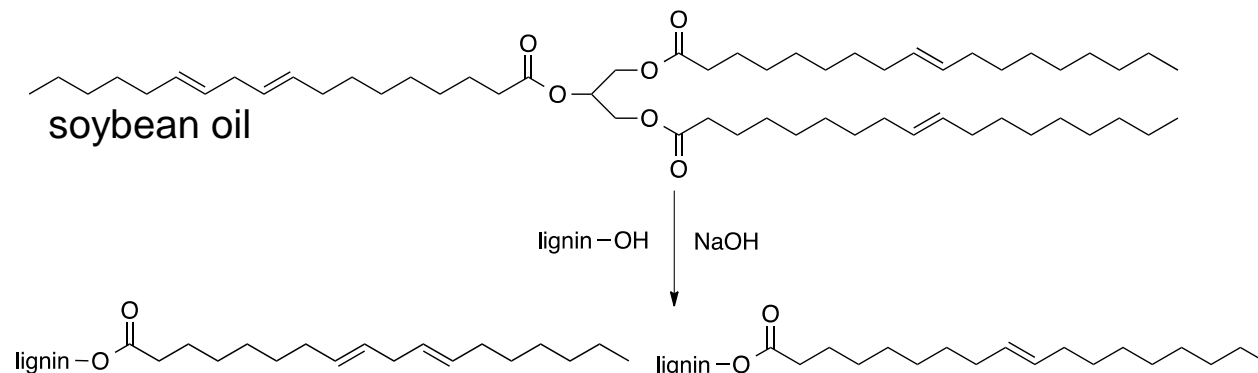
Effects of oleation stoichiometry on particle size and molecular weight

Sample	$n_{L-OH}/n_{M_o}$ (molar ratio)	Particle size ( $\mu m$ )	Molecular weight		
			$M_n$	$M_w$	$M_w/M_n$
milled OL	/	14.4	2650	8140	3.1
oleated OL#1	1:0.5	3.8	1220	3730	1.8
oleated OL#2	1:0.6	2.0	900	2230	2.4
oleated OL#3	1:0.7	3.2	1320	4090	3.1
oleated OL#4	1:0.8	2.6	1290	4310	3.3

## GPC curves of oleated OL

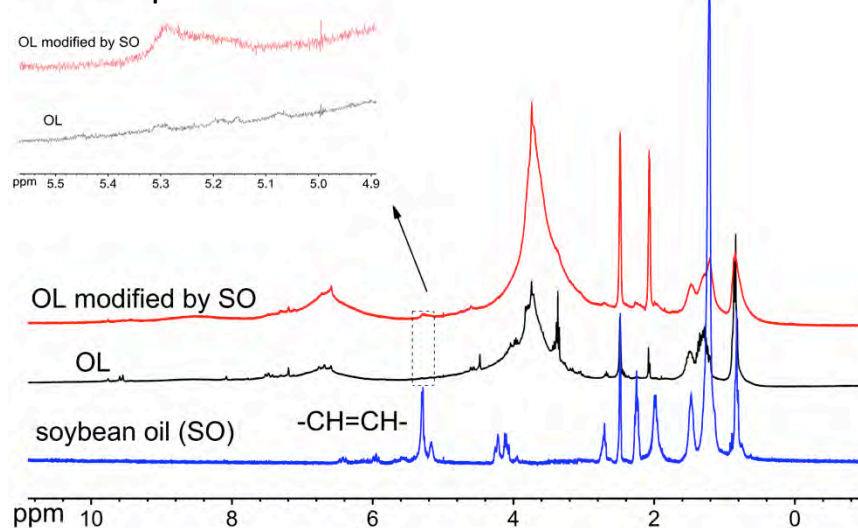


# Transesterification between lignin and fatty oils in ball milling

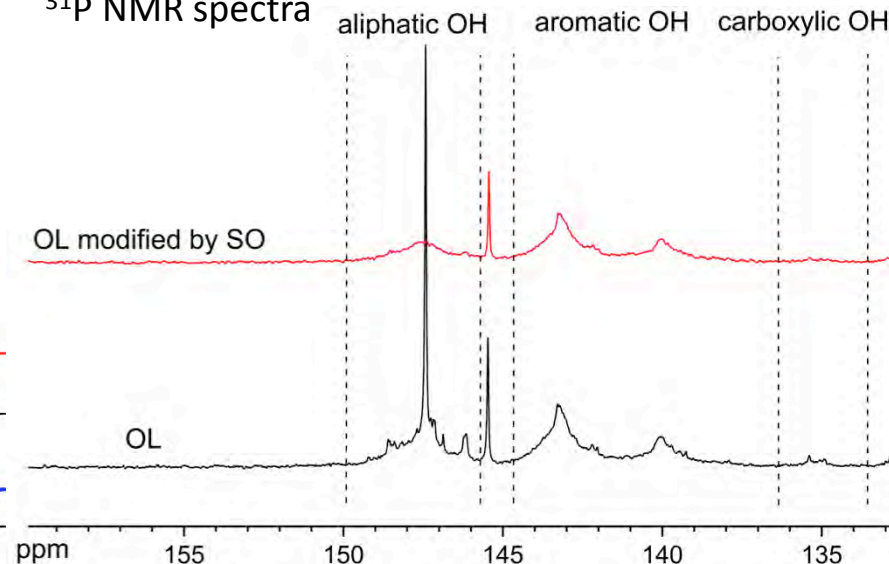


## Identification of chemical structure for modified *organosolv lignin (OL)*

### <sup>1</sup>H NMR spectra



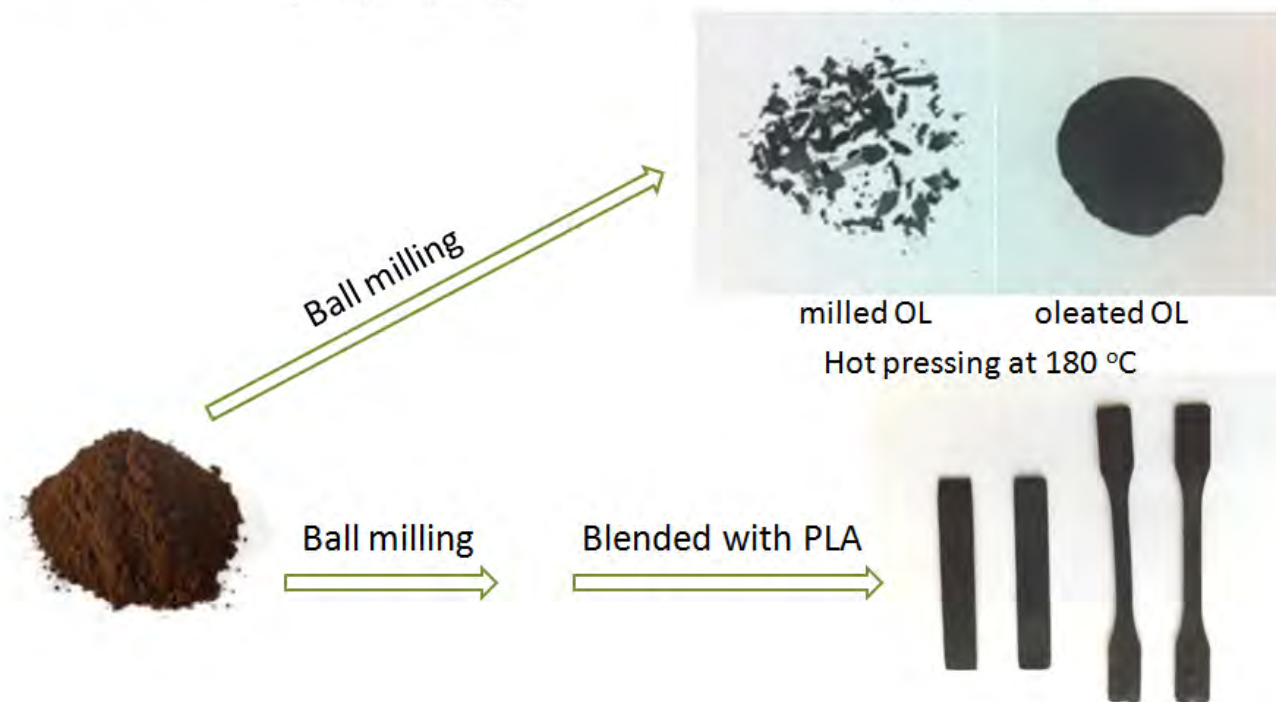
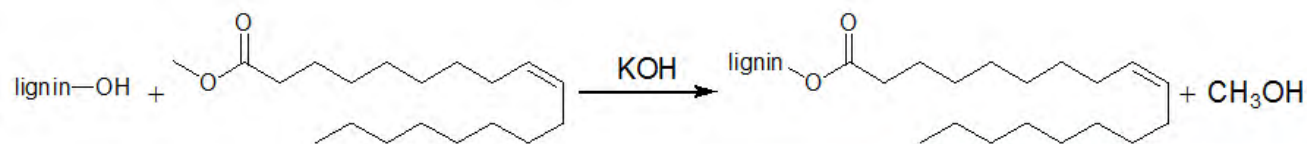
### <sup>31</sup>P NMR spectra



**Aliphatic hydroxyls** 2.57 mmol/g → 1.04 mmol/g

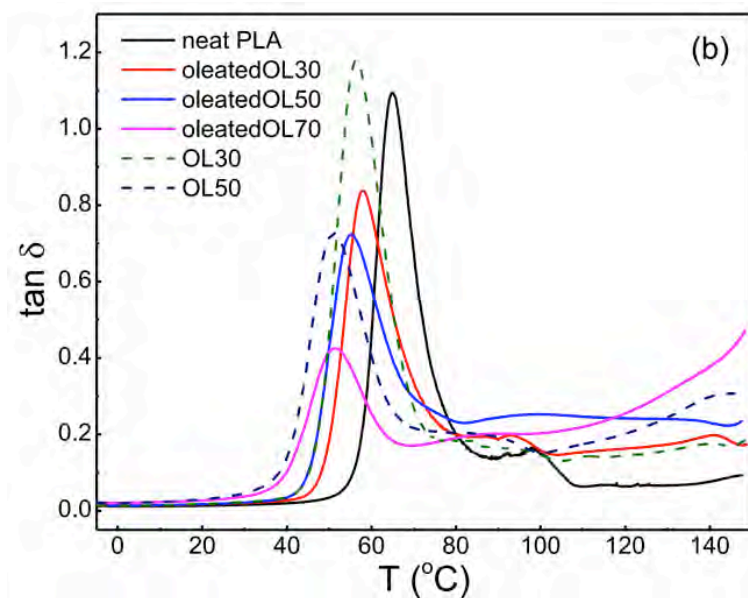
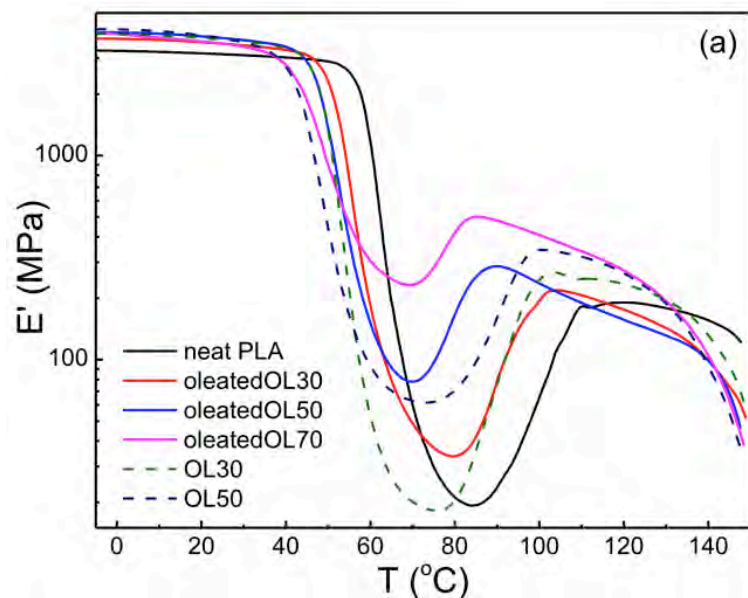
**Aromatic hydroxyls** 2.60 mmol/g → 2.14 mmol/g

# Thermodynamic properties of PLA/lignin blends



**Thermoplasticity of lignin is improved after oleation of lignin**

## DMA curves of (a) storage modulus and (b) $\tan \delta$ of PLA/lignin blends



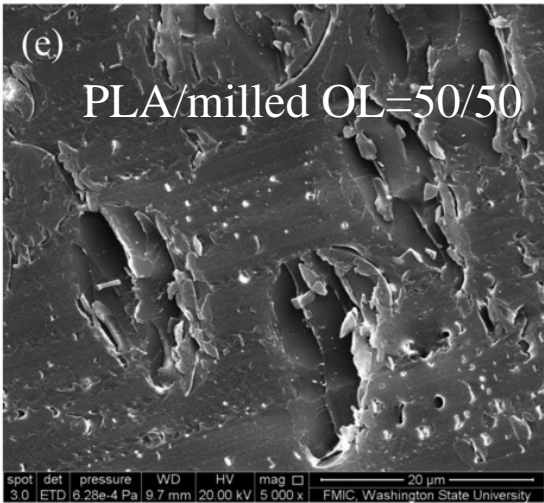
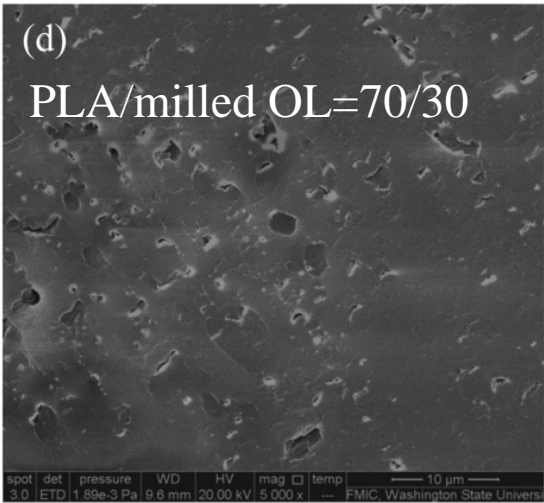
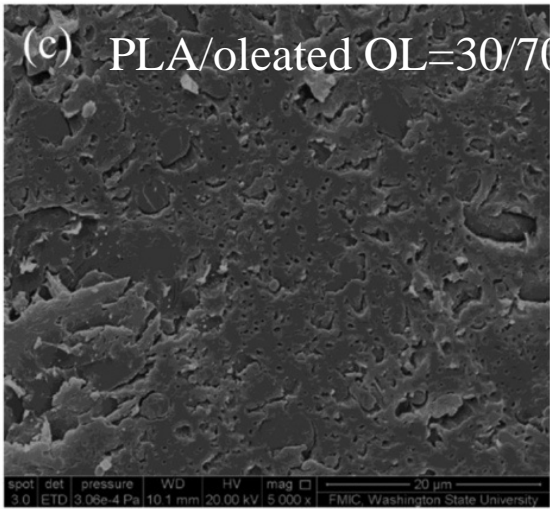
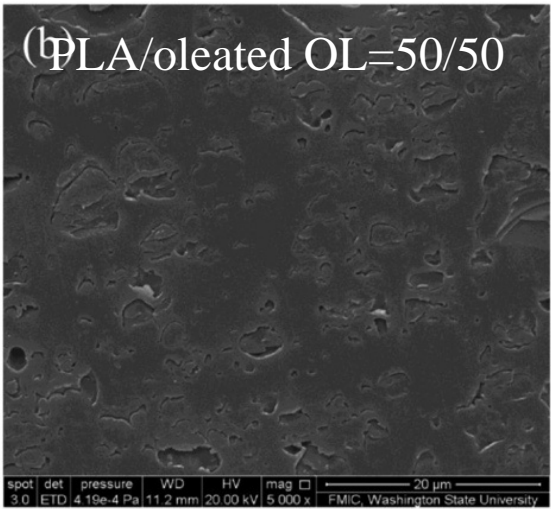
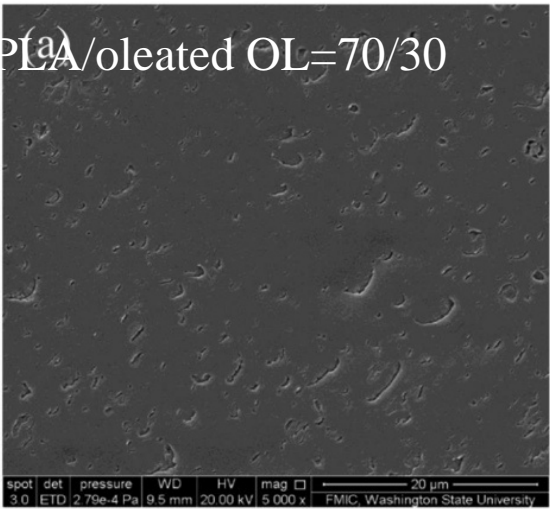
## Thermal properties of PLA and its blends

Sample	$T_g$ (°C) <sup>a</sup>	$T_g$ (°C) <sup>b</sup>	$T_{cc}$ (°C) <sup>b</sup>
neat PLA	64.9	57.2	115.8
oleated OL30	58.0	54.4	98.3
OL30	56.1	52.1	101.8
oleated OL50	55.3	49.8	82.8
OL50	51.1	43.7	95.1
Oleated OL70	51.4	44.8	82.9

a: from DMA  
b: from DSC



SEM micrographs of sliced cross-section surfaces of PLA/lignin blends

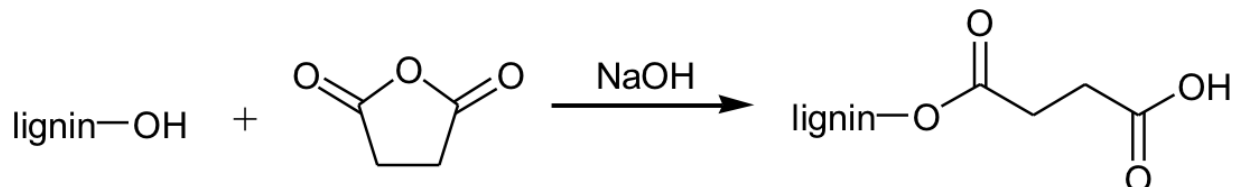


**Tensile properties of PLA blends with OL and oleated OL**

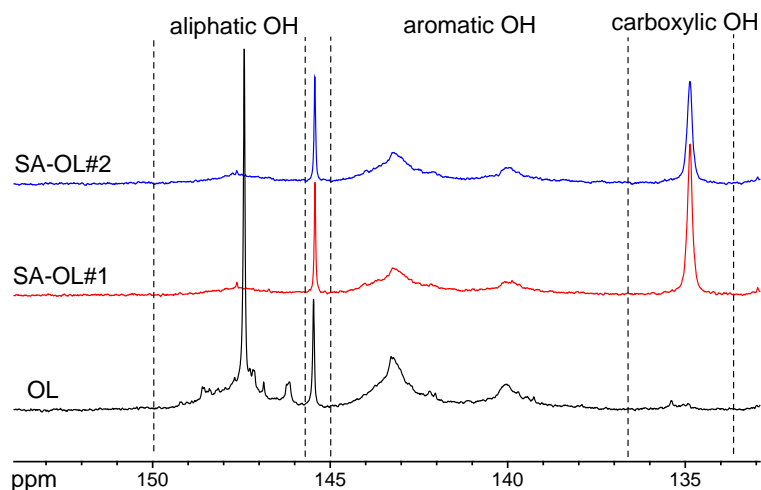
Sample	Elastic modulus GPa	Strength MPa	Elongation @ break %
neat PLA	$3.83 \pm 0.30$	$58.30 \pm 3.39$	$1.48 \pm 0.12$
oleated OL30	$3.71 \pm 0.16$	$50.17 \pm 1.29$	$1.40 \pm 0.02$
milled OL30	$3.63 \pm 0.41$	$37.16 \pm 2.07$	$0.94 \pm 0.06$
oleated OL50	$3.62 \pm 0.20$	$30.36 \pm 0.97$	$0.95 \pm 0.02$
OL50	$3.50 \pm 0.23$	$25.83 \pm 1.34$	$0.75 \pm 0.13$
oleated OL70	$3.29 \pm 0.13$	$11.65 \pm 1.53$	$0.60 \pm 0.03$



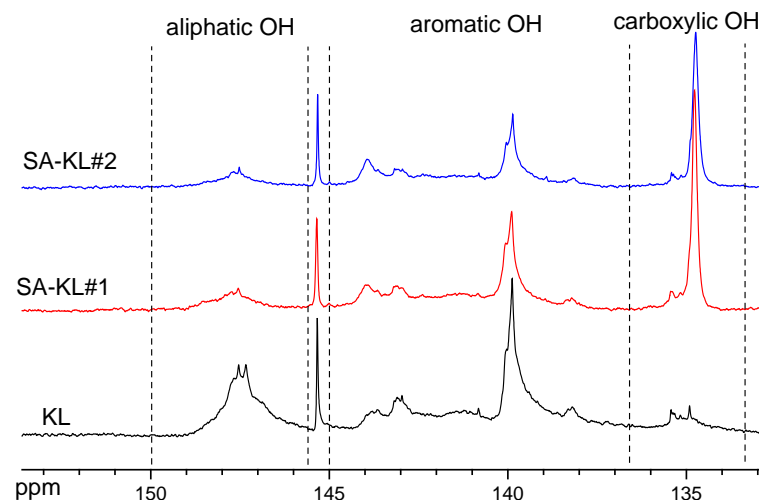
## Identification of chemical structure for modified lignin



### Organosolv lignin



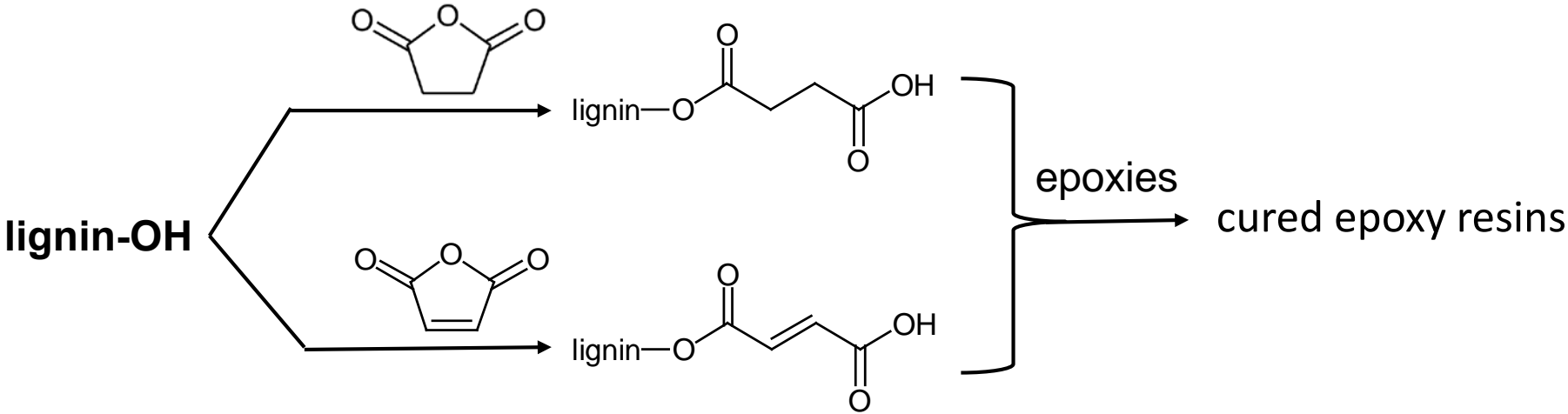
### Kraft lignin



**<sup>31</sup>P NMR spectra**

Effects of esterification stoichiometry on hydroxyl value of modified lignin

Sample	$n_{\text{L-OH}} : n_{\text{SA}}$	Hydroxyl value (mmol/g)		
		Aliphatic OH	Aromatic OH	Carboxylic OH
OL	/	2.57	2.60	0.17
SA-OL#1	1:1	0.24	1.73	1.11
SA-OL#2	1:0.5	0.41	2.02	0.74



# Conclusions

- Lignin can be partially depolymerized to yield low MW oligomers by hydrogenolysis under the catalysis of Raney Ni in alkaline solution of mixed dioxane/H<sub>2</sub>O solvent or base catalyzed depolymerization in methanol under moderate temperature and pressure
- The resulting PDL can be effectively turned into epoxy monomer by reacting with epichlorhydrin. PDL-epoxy cured with the biobased TMA modified asphalt exhibited improved performance
- Mechanochemical process as a green and solvent-free method can be used for the modification of lignin via transesterification. The compatibility of the modified lignin by methyl oleate with PLA was greatly increased
- The novel strongly swellable hydrogel was successfully prepared from lignosulfonate amine and PEGDEG.

# Acknowledgements

## Contributors

- Junna Xin
- Jianglei Qin
- Xiaoxu Teng
- Xiaojie Guo
- Ran Li



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